

A $\text{Ti}_2\text{Ru}_2\text{Pd}_2$ Oxo–Sulfido Cluster Synthesized by Linking Two Rationally Preorganized TiRuPdS_2 Heterotrimetallic Units with an Oxo Ligand: Its Reaction with an Alkyne

Shigeki Kuwata,^{*,†} Shin-ichiro Kabashima,[†]
Youichi Ishii,[†] and Masanobu Hidai^{*,‡}

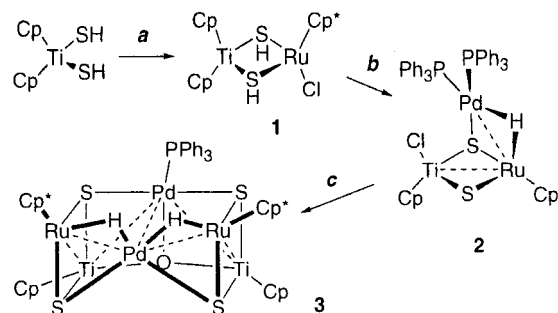
Department of Chemistry and Biotechnology
Graduate School of Engineering, The University of Tokyo
Hongo, Bunkyo-ku, Tokyo 113-8656, Japan
Department of Materials Science and Technology
Faculty of Industrial Science and Technology
Science University of Tokyo, Noda, Chiba 278-8510, Japan

Received November 27, 2000

Bridge-assisted aggregation of preorganized polynuclear cluster complexes provides a rational access to more complicated yet ordered inorganic architectures, which are not obtained merely by the spontaneous self-assembly methods. The structural motifs of the preorganized building blocks used in this stepwise approach are, however, still limited to highly symmetrical ones such as regular triangles,¹ tetranuclear cubanes,² and hexanuclear octahedrons,³ and these building blocks contain two kinds of metals at most. As an extension of our study on the mixed-metal sulfido complexes containing titanium,^{4–7} we report here an oxo-assisted coupling reaction of two TiRuPd heterotrimetallic sulfido fragments having a deformed triangular core, which results in the formation of a novel $\text{Ti}_2\text{Ru}_2\text{Pd}_2\text{S}_4\text{O}$ hexanuclear cluster. Its reactivity toward dimethyl acetylenedicarboxylate (DMAD) is also described.

The starting heterotrimetallic building block $[(\text{CpTiCl})(\text{Cp}^*\text{Ru})\text{Pd}(\text{PPh}_3)_2](\mu_3\text{-S})(\mu_2\text{-S})(\mu_2\text{-H})$ (**2**; Cp = $\eta^5\text{-C}_5\text{H}_5$; Cp* = $\eta^5\text{-C}_5\text{Me}_5$) has been prepared by the reaction of the heterobimetallic hydrosulfido complex $[\text{Cp}_2\text{Ti}(\mu_2\text{-SH})_2\text{RuClCp}^*]$ (**1**)⁵ with $[\text{Pd}(\text{PPh}_3)_4]$ in the presence of triethylamine (Scheme 1).⁷ Cluster **2** has an unprecedented $\text{M}_3(\mu_3\text{-S})(\mu_2\text{-S})$ core with a labile Cl and PPh_3 ligands, which would provide promising handholds for further derivatization including nuclearity expansion. Taking account of the oxophilic nature of titanium, we have here treated **2** with water. When an excess of water was added to a THF solution of the trinuclear cluster **2** in the presence of a base, the

Scheme 1^a



^a Reagents and conditions: (a) $[(\text{Cp}^*\text{Ru})_4(\mu_3\text{-Cl})_4]$, rt; (b) (i) Et_3N , -78°C ; (ii) $[\text{Pd}(\text{PPh}_3)_4]$, -78°C to rt; (c) H_2O (20 equiv), LiBHET_3 (5 equiv), -78°C to rt.

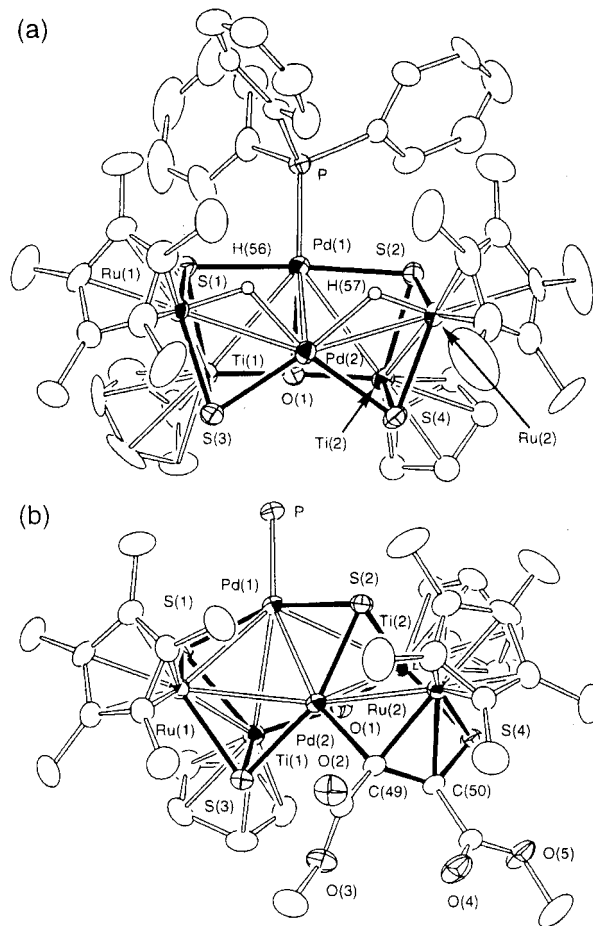


Figure 1. Structures of **3** (a) and **4** (b). Thermal ellipsoids are shown at the 30% probability level. The phenyl groups in **4** as well as hydrogen atoms except for the hydrido ligands in **3** are omitted for clarity.

hexanuclear cluster $[(\text{CpTi})_2(\text{Cp}^*\text{Ru})_2\text{Pd}_2(\text{PPh}_3)_4(\mu_3\text{-S})_4(\mu_3\text{-O})(\mu_2\text{-H})_2]$ (**3**) has been obtained (Scheme 1).^{8,9} An X-ray analysis of **3** has confirmed the $\text{Ti}_2\text{Ru}_2\text{Pd}_2\text{S}_4\text{O}$ core, in which two corner-voided $\text{TiRuPd}_2\text{S}_2\text{O}$ cuboidal fragments share the Pd_2O face, a pseudo-mirror plane of this molecule (Figure 1a).¹⁰ Each of the $\text{Pd}(2)\text{—Ru}$ edges in the final difference Fourier map. These hydrido ligands as well

(8) Full experimental procedures are given in the Supporting Information.
(9) The best yield (58%) was achieved by using 5 equiv of LiBHET_3 as a base. Use of other bases such as LiOH and $\text{NaN}(\text{SiMe}_3)_2$ also gave **3**, although the yields were lower.

[†] The University of Tokyo.

[‡] Science University of Tokyo.

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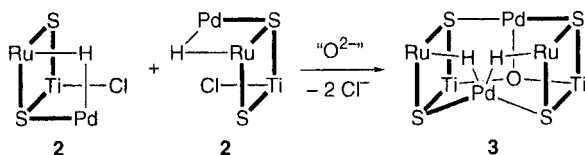
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Scheme 2



as the S(3) and S(4) atoms constitute the distorted square-planar coordination geometry around the Pd(2) atom. The Pd(1) atom also lies in a square-planar environment with the S(1), S(2), P, and O(1) atoms, although the considerably long Pd(1)–O(1) distance (2.323(6) Å) suggests that the interaction between these atoms is weak. Judging from the metal–metal distances, **3** has seven metal–metal bonds, which are indicated as dashed lines in Scheme 1. It should be mentioned that a related incomplete double cubane-type heterobimetallic cluster [(CpTi)₂Rh₄(CO)₆-(μ₄-O)(μ₃-S)₄] has been prepared by spontaneous self-assembly of monotitanium and dirhodium fragments.¹¹ The O atom in this Ti₂Rh₄ cluster is bound to four metal atoms (two Ti and two Rh atoms) unlike that in **3**.

Formation of **3** is featured by the multiple linkage of two molecules of **2** by different types of arrays of elements, that is, the hard (Ti–O–Ti) and soft (Ru–S–Pd) atoms (Scheme 2).

This is in marked contrast with the reaction of [CpTiCl(SCH₂-CH₂CH₂S)] with water in the presence of imidazole, which results in the linkage of the monotitanium fragment only by an oxo ligand.¹²

Cluster **3** has an intriguing hydridopalladium center surrounded by sterically demanding Cp* and PPh₃ ligands. Our continuing interest in the reactivities of small molecules on such noble metal centers embedded in polynuclear metal–sulfur cores^{4,13} led us to examine the reaction of **3** with DMAD, which afforded the thiadimetallacycle cluster **4** in 36% yield (eq 1).⁸

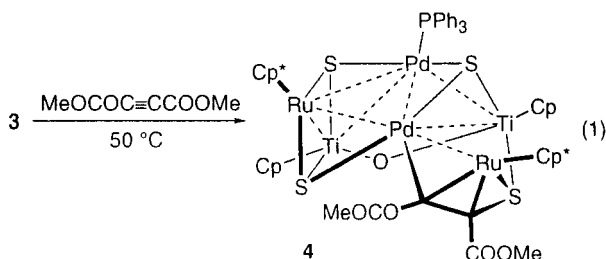


Figure 1b depicts the X-ray structure of **4**.¹⁰ A DMAD molecule has formally been inserted into one of the Pd–S bonds in **3** to give an almost planar, five-membered thiapalladitanacycle,

(10) Crystal data for **3**·THF·MeOH: monoclinic, *Cc* (No. 9), *a* = 23.873(5) Å, *b* = 11.955(4) Å, *c* = 22.234(4) Å, β = 116.47(1)°, *V* = 5680(2) Å³, *T* = 294 K, *R* (*R*_w) = 0.041 (0.047), GOF = 1.54. For **4**·Et₂O: orthorhombic, *Pbca* (No. 61), *a* = 25.129(4) Å, *b* = 25.477(4) Å, *c* = 18.786(3) Å, *V* = 12027(3) Å³, *T* = 294 K, *R* (*R*_w) = 0.048 (0.048), GOF = 1.46.

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which is further coordinated to the Ru(2) atom. The insertion has also led to the formation of the Ti(2)–Pd(2), Ru(1)–Pd(1), and Pd(2)–S(2) bonds, whereas the Ti(2)–Ru(2), Pd(2)–S(4), Ru(2)–S(2), and Pd(1)–O(1) bonds in **3** have now been cleaved. The ¹H NMR spectrum of **4** no longer exhibits any signals ascribed to hydrido protons as well as vinyl and hydrosulfido ones. The electrospray ionization (ESI) mass and IR spectra of **4** also suggest the absence of hydrido and hydrosulfido ligands in **4**. We thus concluded that the hydrido hydrogen atoms in **3** have been lost during the reaction, although we could not detect dihydrogen gas nor hydrogenated organic compounds such as dimethyl fumarate.

The reaction of a μ₃-sulfido ligand and an alkyne is quite scarce. We have already demonstrated that the triruthenium sulfido cluster [(Cp*₃Ru)₃(μ₃-S)(μ₃-Cl)] reacts with MeC≡CCOOMe at 50 °C to afford a dinuclear thiametallacycle complex,¹⁴ although the trinuclear core has been ruptured, and the detailed reaction mechanism is unclear. More prevalent is addition of alkynes to μ₂-bridging or terminal sulfido ligands.^{15,16} Terminal sulfido complexes undergo the [2 + 2] cycloaddition to afford thiametallacyclobutenes¹⁷ in addition to the [2 + 3] cyclization leading to dithiolene complexes;¹⁸ in any case, the insertion of an alkyne into a M–(μ₃-S) bond is unprecedented. As to other sulfur-based ligands, side-on disulfido complexes are known to react with alkynes to afford such insertion products, namely, five-membered MSSC(R)=CR dithiametallacycles,¹⁹ as well as dithiolene complexes. Insertion of alkynes to the Pd–SR bond has also been proposed as a key step in Pd-catalyzed additions of diorganodisulfides RSSR and thiols RSH to the alkynes.²⁰

In summary, the novel hexanuclear oxo–sulfido cluster **3** with three very distinct metals has been synthesized in a stepwise and rational manner as outlined in Scheme 1. Cluster **3** further undergoes the unprecedented insertion of an alkyne into a Pd–(μ₃-S) bond without degradation of the hexanuclear framework.

Acknowledgment. We thank Dr. Hidetake Seino (Institute of Industrial Science, The University of Tokyo) for ESI mass analysis of **3** and **4**. Financial support by the Ministry of Education, Science, Sports, and Culture of Japan (Grant Nos. 09102004 and 12750755) and The Foundation “Hattori Hokokai” (to S.K.) is appreciated.

Supporting Information Available: Experimental details for the preparation of **3** and **4** (PDF). An X-ray crystallographic file (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

JA0040666

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